

Table S1. Volumes of reactants and products in reactions 1 to 4, calculated from coefficients of the balanced reactions and from molar volumes taken from Helgeson et al. (1978; with exception for H₂O at 800 °C and 0.8 GPa taken from Burnham et al., 1969). Mineral abbreviations as in Figure 4.

	Reactants							Products					
	Ol	Pl	Cpx	Spl	Grt	Opx	H ₂ O	sum	Opx	Cpx	Spl	Grt	Amp
V(cm ³ /m)	43.79	100.79	66.09	39.71	148.47	62.56	21.99		62.56	66.09	39.71	148.47	272.92
Reaction 1	0.70	0.36							0.34	0.37	0.35		
Volume (cm ³)	30.65	36.28					66.94	21.27	24.45	13.90			59.62
%Volume	45.79	54.21						35.68	41.02	23.31			
Reaction 2			0.69	0.52							0.51		
Volume (cm ³)			45.60	20.65			66.25				75.72		75.72
%Volume			68.83	31.17							100.00		
Reaction 3			0.02	0.89	0.07	0.04	0.05			0.45		0.05	
Volume (cm ³)			1.32	35.34	10.39	2.50	1.10	50.66		17.87		13.65	31.52
%Volume			2.61	69.76	20.52	4.94	2.17			56.69		43.29	
Reaction 4	0.70	0.42						0.28	0.37	0.32	0.09		
Volume (cm ³)	30.65	42.33					72.98	17.52	24.45	12.71	13.36		68.04
%Volume	42.00	58.00						25.74	35.94	18.68	19.64		